

## A model for the temperature effect on onset velocity of turbulent fluidization of Geldart type A particles

Jeong-Hoo Choi<sup>\*†</sup>, Ho-Jung Ryu<sup>\*\*</sup>, and Chang-Keun Yi<sup>\*\*</sup>

<sup>\*</sup>Department of Chemical Engineering, Konkuk University, Seoul 143-701, Korea

<sup>\*\*</sup>Korea Institute of Energy Research, Daejeon 305-343, Korea

(Received 14 March 2010 • accepted 29 May 2010)

**Abstract**—This study proposes a model to predict the temperature effect on the onset velocity of turbulent fluidization ( $u_c$ ) of Geldart type A particles. It was found that void splitting could occur in the whole bed when the initial bubble size at the distributor ( $d_{bo}$ ) was greater than the maximum stable bubble size ( $d_{bmax}$ ) or the equilibrium bubble size ( $d_{beq}$ ). The proposed model was successful to fit the trend of temperature effect on the  $u_c$ . When the bubble growth was limited by the  $d_{bmax}$ , the  $u_c$  increased with an increase of temperature. However, when the  $d_{beq}$  was smaller than  $d_{bmax}$ , the  $u_c$  decreased with an increase of temperature because the minimum fluidizing velocity and thus the  $d_{beq}$  decreased. The  $u_c$  increased initially and decreased later when the  $d_{beq}/d_{bmax}$  starting with the ratio  $>1$  decreased to the ratio  $<1$  as temperature increased. The present model predicted the trend properly that any other existing correlations could not be able to reflect.

Key words: Model, Temperature Effect, Transition Velocity, Onset Velocity, Turbulent Fluidized Bed

### INTRODUCTION

The turbulent fluidization regime is commonly considered to lie between bubbling fluidization and the fast fluidization regime. The turbulent regime is known to result from disappearance of large voids by predominance of void splitting over void coalescence. Turbulent fluidization is commonly utilized in industrial fluidized-bed reactors due to vigorous gas-solids contacting, favorable bed-to-surface heat transfer, high solids hold-ups, and limited axial mixing of gas [1].

Although extensive results have been reported to quantify the onset velocity of turbulent fluidization ( $u_c$ ), the effect of temperature on it is not clear yet for Geldart type A particles [2,3]. The  $u_c$  increased with temperature for Geldart type A and B particles in the study of Cai et al. [4,5]. However, they measured the static pressure in the middle of bed height that was improper to reflect the property of the whole bed. Chehbouni et al. [2] reported that the  $u_c$  decreased with an increase of temperature in the bed of FCC particles (Geldart type A; 78 μm, 1,450 kg/m<sup>3</sup>). These results were related to the effect of temperature on bubble size, which was significantly different for fine and large particle diameters. Moreover, no published correlation could predict adequately those experimental data. The  $u_c$  in the bed of sand particles (Geldart type B; 250 μm, 2,560 kg/m<sup>3</sup>) increased with an increase of temperature. According to the study of Peeler et al. [3], as temperature increased, the  $u_c$  decreased after an initial increase in the bed of sand particles (Geldart type B; 130 μm, 4,400 kg/m<sup>3</sup>) using N<sub>2</sub> as fluidizing gas and decreased in the bed of alumina particles (Geldart type A; 70 μm, 2,800 kg/m<sup>3</sup>) using N<sub>2</sub> or He as fluidizing gas. None of the reported correlations for estimating  $u_c$  satisfactorily predicts this behavior.

This study is to propose a model for temperature effect on transition velocity to turbulent fluidization ( $u_c$ ) of Geldart type A particles. Two cases of void splitting were considered for the fluidized bed of Geldart group A and AB particles. One is that the equilibrium bubble diameter is smaller than the maximum stable bubble diameter. Then the  $u_c$  decreased with an increase of temperature. Another is that the equilibrium bubble diameter is greater than the maximum stable bubble diameter. Then the  $u_c$  increased with temperature.

### MODEL

The void size in a bubbling or slugging bed is limited by the maximum stable bubble size according to the study of Harrison et al. [6]. The bubble size in the bed of Geldart type A particles is controlled additionally by equilibrium bubble size because bubble splitting occurs [7]. Therefore, bubble breakup could occur in the whole bed when the initial bubble size at the distributor is greater than the smaller one of the maximum stable bubble size and the equilibrium bubble size.

At the usual gas velocity of turbulent bed including the bed of Chehbouni et al. [2] and Peeler et al. [3], which were considered in this study, the bubble diameter formed initially at the distributor is bigger than the pitch of the distributor nozzle so given by the following correlation for the porous plate distributor [8,9].

$$d_{bo} = \frac{3.685(u - u_{mf})^2}{g} \quad (1)$$

Harrison et al. [6] proposed the following relationship for the maximum stable bubble size:

$$\frac{d_{bmax}}{d_p} = 1.32 \frac{\rho_p - \rho_g}{\rho_g} \frac{\frac{\rho_p}{\rho_g} - \epsilon_{mf}}{1 - \epsilon_{mf}} \quad (2)$$

<sup>†</sup>To whom correspondence should be addressed.  
E-mail: choijhoo@konkuk.ac.kr

Since  $\frac{\rho_p - \rho_g}{1 - \varepsilon_{mf}} - \varepsilon_{mf}$  is nearly unity in most gas fluidized beds, the simplified form gives

$$d_{b\ max} = 1.32 \frac{\rho_p - \rho_g}{\rho_g} d_p \quad (3)$$

When this relationship is applied to the initial bubble size (Eq. (1)) formed at the distributor as a criterion to begin bubble breakup throughout the bed, the onset velocity of turbulent fluidization  $u_c$  is

$$u_c = u_{mf} + 0.5985 \left( \frac{d_p(\rho_p - \rho_g)g}{\rho_g} \right)^{0.5} \quad (4a)$$

$$\frac{d_p u_c \rho_g}{\mu} = \frac{d_p u_{mf} \rho_g}{\mu} + 0.5985 \left( \frac{d_p^3 \rho_g (\rho_p - \rho_g) g}{\mu^2} \right)^{0.5} \quad (4b)$$

$$Re_c = Re_{mf} + 0.5985 Ar^{0.5} \quad (4c)$$

The  $Re_{mf}$  can be given by the correlation of Wen and Yu [10]:

$$Re_{mf} = [(33.7)^2 + 0.0408 Ar]^{0.5} - 33.7 \quad (5)$$

Then, the onset condition of turbulent fluidization is in another form:

$$Re_c = [(33.7)^2 + 0.0408 Ar]^{0.5} - 33.7 + 0.5985 Ar^{0.5} \quad (6)$$

When the bubble growth is limited by the maximum stable bubble size, the onset velocity of turbulent fluidization  $u_c$  increases with an increase of temperature as can be seen in Eq. (4a).

The equilibrium bubble size can be given by following correlations of Choi et al. [7]:

$$d_{beg} = \frac{6.792 k (u - u_{mf})}{f_s} \quad (7)$$

$$k = \frac{u - (u/u_{mf})^{0.62} u_{mf}}{u - u_{mf}} \quad (8)$$

$$f_s^* = 6.47 \times 10^{-4} \left( \frac{u}{u_{mf}} \right)^{0.454} \frac{g}{u_{mf}} \quad (9)$$

The  $k$  is bubble flow fraction of excess gas velocity and the  $f_s^*$  splitting frequency of a single bubble. The equilibrium bubble size is considered now that the bubble splitting is important in the bed of Geldart type A particles. When Eq. (7) is applied to the initial bubble size (Eq. (1)) formed at the distributor as a criterion to start the bubble breakup throughout the bed, the onset velocity of turbulent fluidization is

$$\left( \frac{u_c}{u_{mf}} \right)^{2.454} - 2 \left( \frac{u_c}{u_{mf}} \right)^{1.454} - 2849 \left( \frac{u_c}{u_{mf}} \right) + 2849 \left( \frac{u_c}{u_{mf}} \right)^{0.62} + \left( \frac{u_c}{u_{mf}} \right)^{0.454} = 0 \quad (10)$$

The solution is

$$\frac{u_c}{u_{mf}} = 217.5 \quad (11)$$

Therefore, if the bubble growth is limited by the equilibrium bubble size, the  $u_c$  decreases with an increase of temperature because the minimum fluidizing velocity and thus the equilibrium bubble size decrease.

Combining Eqs. (3) and (7) to (9) gives

$$\frac{d_{beg}}{d_{b\ max}} = 7953 \frac{Re_{mf}^2}{Ar} \left\{ \frac{\left( \frac{u}{u_{mf}} \right) - \left( \frac{u}{u_{mf}} \right)^{0.62}}{\left( \frac{u}{u_{mf}} \right)^{0.454}} \right\} \quad (12)$$

First, the bubble size in the bed of Geldart type A, B and D particles is limited by the maximum stable bubble size. The bubble size in the bed of Geldart type A particles is controlled additionally by the equilibrium bubble size because the bubble splitting is important. The bubble splitting frequency in the bed of Geldart type B and D particles is so low that the equilibrium bubble size is greater than the stable bubble size and thus no real limitation to the bubble growth. Therefore, bubble breakup occurs in the entire bed if the initial bubble size formed at the distributor is greater than the smaller one of the maximum stable bubble size and the equilibrium bubble size.

## RESULTS AND DISCUSSION

Fig. 1 shows the ratio of the equilibrium bubble diameter to the maximum stable bubble diameter,  $d_{beg}/d_{bmax}$  calculated by Eq. (12) at  $u=u_c$ , which was measured by Chehbouni et al. [2] and Peeler et al. [3]. The  $d_{beg}/d_{bmax}$  for the bed of sand and FCC particles was smaller or greater than unity, whereas for the bed of alumina particles smaller than unity. Therefore, this result confirms that bubble breakup in the bed could be caused by the initial bubble size formed at the distributor greater than the maximum stable bubble size or the equilibrium bubble size. The present discussion did not consider the data of Chehbouni et al. [2] obtained from sand beds because the  $d_{beg}/d_{bmax} > 1$  for those beds. The bubble splitting frequency in the bed of Geldart type B and D particles is so low that the equilibrium bubble size is greater than the stable bubble size and thus no real limitation to the bubble growth.

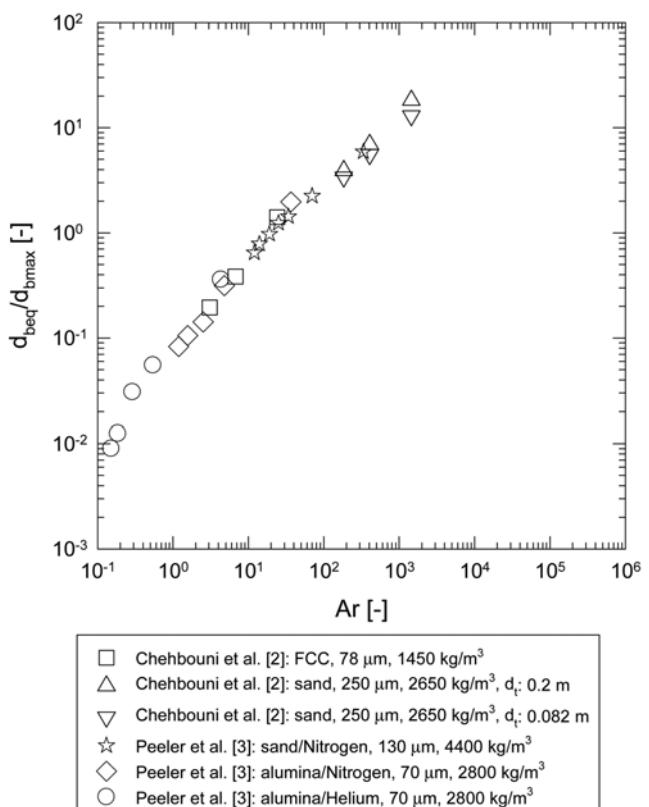


Fig. 1.  $d_{beg}/d_{bmax}$  versus  $Ar$  at  $u_c$ .

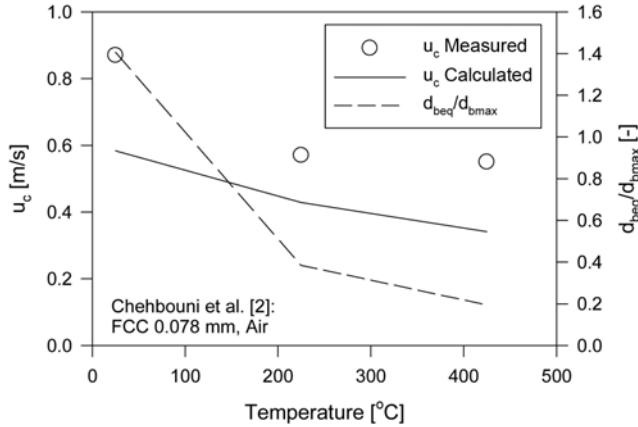


Fig. 2. Effect of temperature on  $u_c$  and  $d_{beq}/d_{bmax}$  (Chehbouni et al. [2]; FCC 0.078 mm, air).

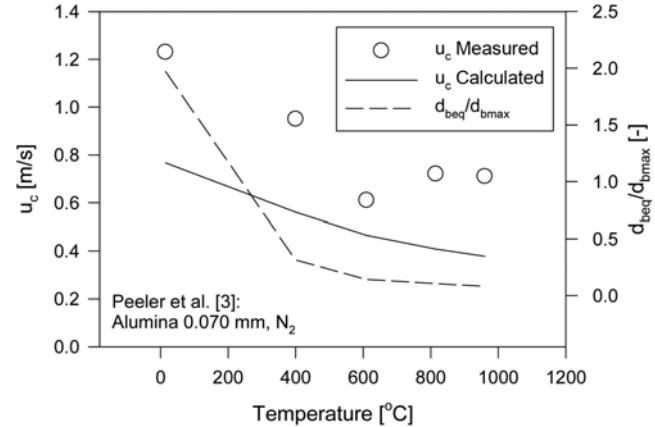


Fig. 4. Effect of temperature on  $u_c$  and  $d_{beq}/d_{bmax}$  (Peeler et al. [3]; alumina 0.070 mm,  $N_2$ ).

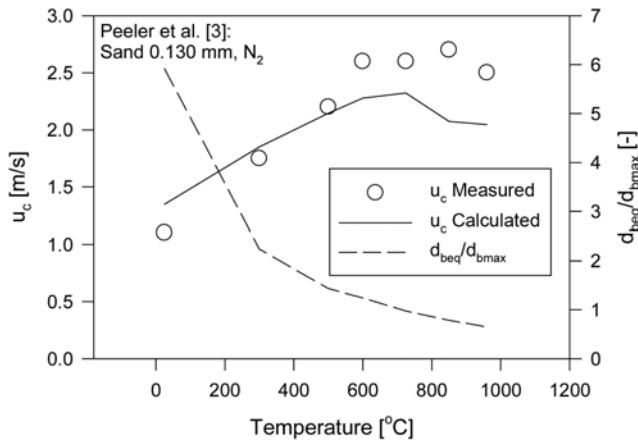


Fig. 3. Effect of temperature on  $u_c$  and  $d_{beq}/d_{bmax}$  (Peeler et al. [3]; sand 0.130 mm,  $N_2$ ).

Fig. 2 shows the comparison of the present model with data measured by Chehbouni et al. [2]. The maximum stable bubble diameter is greater than the equilibrium bubble diameter at temperature > 104 °C. So the  $u_c$  was determined by Eq. 4(a) for temperature < 104 °C and Eq. (11) for temperature > 104 °C. The present model could predict the temperature effect on the  $u_c$  reasonably well that decreased with an increase of temperature. However, the equilibrium diameter seemed to be somewhat underestimated so that the present model underestimated the  $u_c$ .

Fig. 3 shows a comparison of the present model with the data of Peeler et al. [3], which were measured in sand beds using  $N_2$  gas as fluidizing gas. The  $u_c$  initially increased but decreased later as temperature increased. The  $d_{beq}/d_{bmax}$  was predicted greater than unity at temperature < 725 °C so the  $u_c$  was determined by Eq. (4a). However, the  $d_{beq}/d_{bmax}$  was predicted smaller than unity at temperature > 725 °C so Eq. (11) determined the  $u_c$ . The present model could predict the measured trend well. It was the trend that any other existing correlations could not follow reasonably [3].

Fig. 4 compares the present model with the data of Peeler et al. [3], measured in the alumina bed using  $N_2$  gas as fluidizing gas. The  $d_{beq}/d_{bmax}$  was predicted greater than unity at temperature < 238 °C so the  $u_c$  was determined by Eq. (4a). However, the  $d_{beq}/d_{bmax}$  was

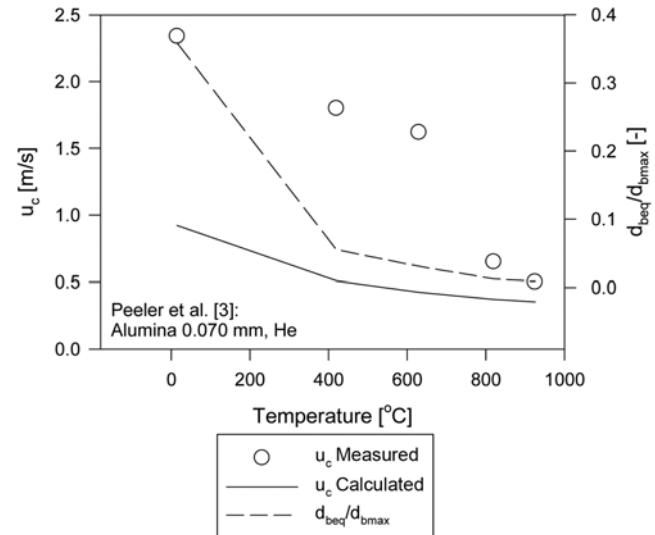


Fig. 5. Effect of temperature on  $u_c$  and  $d_{beq}/d_{bmax}$  (Peeler et al. [3]; alumina 0.070 mm, He).

predicted smaller than unity at temperature > 238 °C so Eq. (11) determined the  $u_c$ . So the  $u_c$  looked just decreasing with an increase of temperature. The present model could predict the measured trend well at the same condition.

Fig. 5 compares the present model with the data of Peeler et al. [3], measured in the alumina bed using He gas as fluidizing gas. The  $d_{beq}/d_{bmax}$  was predicted smaller than unity within the tested temperature range. The  $u_c$  decreased with an increase of temperature. The present model predicted the trend properly. But it underestimated the  $u_c$  considerably at temperatures 15 °C, 420 °C and 630 °C and could not follow the unlikely big decrease (0.97 m/s) between 630 °C and 820 °C.

Fig. 6 shows the relationship between  $u_c$  and  $u_{mf}$  when  $d_{beq} < d_{bmax}$ . Eq. (11) was in reasonable agreement with the data of Chehbouni et al. [2] and Peeler et al. [3], except for some data of the alumina-He system at 15 °C, 420 °C, and 630 °C. The present knowledge hardly seemed to explain the reason for the disagreement, only the present model seemed to underestimate the  $u_c$  because the equilibrium diameter was underestimated. It is evident that further study

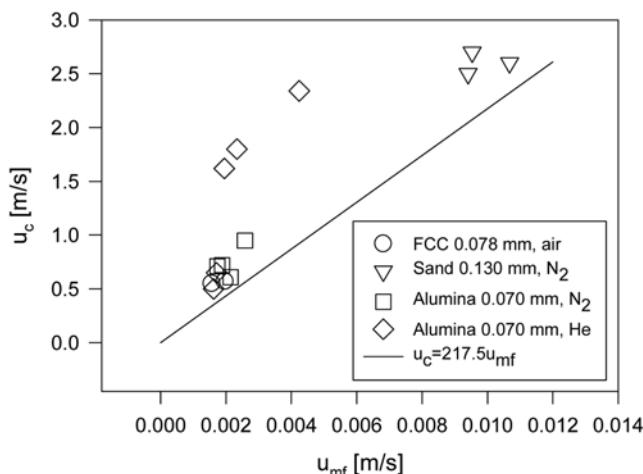


Fig. 6.  $u_c$  versus  $u_{mf}$  [2,3].

is needed to improve the model accuracy.

## CONCLUSIONS

This study proposes a model to predict the temperature effect on the  $u_c$  of Geldart type A particles. The present study found that bubble breakup in the whole bed could be caused by the  $d_{bo} > d_{bmax}$  or  $d_{beg}$ . The present model was successful to fit the trend of temperature effect on the  $u_c$ . When the bubble growth was limited by the  $d_{bmax}$ , the  $u_c$  increased with an increase of temperature. However, when  $d_{beg} < d_{bmax}$ , the  $u_c$  decreased with an increase of temperature because the minimum fluidizing velocity and thus the equilibrium bubble size decreased. The  $u_c$  increased initially and decreased later when the  $d_{beg}/d_{bmax}$  starting with a ratio  $> 1$  decreased to a ratio  $< 1$  as temperature increased. The present model predicted the trend properly that any other existing correlations could not reflect. The present model seemed to underestimate the  $u_c$  because the  $d_{beg}$  was underestimated. Further study is needed to improve the model accuracy.

## ACKNOWLEDGEMENT

This work was supported by the Energy Efficiency & Resources of the Korea Institute of Energy Technology Evaluation and Planning (KETEP) grant funded by the Korea government Ministry of Knowledge Economy (2008CCD11P030000).

## NOMENCLATURE

$$Ar : \text{Archimedes number}, \frac{d_p^3 \rho_g (\rho_p - \rho_g) g}{\mu^2}$$

$d_{beg}$	: equilibrium bubble diameter [m]
$d_{bmax}$	: maximum stable bubble diameter [m]
$d_{bo}$	: initial bubble diameter formed at the distributor [m]
$d_p$	: particle diameter [m]
$d_t$	: column diameter [m]
$f_s^*$	: splitting frequency of a single bubble [1/s]
$g$	: gravitational acceleration, 9.8 [m/s <sup>2</sup> ]
$k$	: bubble flow fraction of excess gas velocity ( $u - u_{mf}$ ) [-]
$Re_c$	: particle Reynolds number at the onset condition of turbulent fluidization, $(d_p u_c \rho_g)/\mu$ [-]
$Re_{mf}$	: particle Reynolds number at minimum fluidization, $(d_p u_{mf} \rho_g)/\mu$ [-]
$u$	: gas velocity [m/s]
$u_c$	: onset velocity of turbulent fluidization [m/s]
$u_{mf}$	: minimum fluidizing velocity [m/s]
$\varepsilon_{mf}$	: bed voidage at minimum fluidization condition [-]
$\rho_g$	: gas density [kg/m <sup>3</sup> ]
$\rho_p$	: solid density [kg/m <sup>3</sup> ]
$\mu$	: gas viscosity [Pa·s]

## REFERENCES

- H. T. Bi, N. Ellis, I. A. Abba and J. R. Grace, *Chem. Eng. Sci.*, **55**, 4789 (2000).
- A. Chehbouni, J. Chaouki, C. Guy and D. Klvana, "Fluidization VIII" Edited by J. F. Large and C. Lagu're, Engineering Foundation, New York, 149 (1995).
- P. K. Peeler, K. S. Lim and R. C. Close, "Circulating Fluidized Bed Technology VI," Edited by J. Werther, **1**, 125 (1999).
- P. Cai, S. P. Chen, Y. Jin, Z. Q. Yu and Z. W. Wang, *AIChE Symposium Series*, **85**(270), 37 (1989).
- P. Cai, Y. Jin, Z. Q. Yu and L. S. Fan, *Ind. Eng. Chem. Res.*, **31**, 632 (1992).
- D. Harrison, J. F. Davidson and J. W. de Kock, *Trans. Instn. Chem. Eng.*, **39**, 202 (1961).
- J. H. Choi, J. E. Son and S. D. Kim, *Ind. Eng. Chem. Res.*, **37**, 2559 (1998).
- D. Kunii and O. Levenspiel, *Fluidization Engineering*, 2<sup>nd</sup> Ed., Butterworth-Heinemann, Boston, 130 (1991).
- K. Miwa, S. Mori, T. Kato and I. Muchi, *Int. Chem. Eng.*, **12**, 187 (1972).
- C. Y. Wen and Y. H. Yu, *AIChE J.*, **12**, 610 (1966).