

Effects of oxygen plasma generated in magnetron sputtering of ruthenium oxide on pentacene thin film transistors

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Abstract—Effects of oxygen plasma generated in a sputtering process for deposition of electrodes on pentacene thin films to configure top-contact (TC) transistors have been thoroughly investigated. Reactive oxygen species severely degraded electrical properties of pentacene films during the deposition of RuO_x electrodes, leading to a failure of devices. In the off-region, the leakage current increased by about two orders of magnitude, and the subthreshold slope also increased by 6.5 times. The top surface of pentacene films was oxidized by oxygen plasma and C-O and C=O bonds were created. The pentacenequinone derivative was confirmed by X-ray photoelectron spectroscopy. The oxidation of pentacene films gives rise to charge traps at the pentacene/electrode interface, which produces a leakage channel between source and drain electrodes. We believe that this side effect of oxygen plasma on the fabrication of TC-devices should be considered carefully.

Keywords: Pentacene, Thin-film Transistor, Ruthenium Oxide, Oxygen, Plasma

INTRODUCTION

Numerous studies on organic semiconductors for several decades have contributed immensely to development of the mobile and display industry. Outstanding luminescent efficiency and mechanical flexibility of this class of materials have been utilized for large-area and flexible organic light-emitting diodes (OLEDs) [1]. Meanwhile, they have been still undesirable for electronic devices requiring high switching speeds due to their inferior charge carrier mobility than conventional inorganic semiconductors such as GaAs and Si. As such, many efforts to enhance the performance of organic thin films transistors (OTFTs) have been made with two perspectives: Improvement of electrical properties of organic semiconductors and optimization of device architecture. For the latter, the deposition order of components (e.g., semiconductors and electrodes) would be crucial since it produces different results of device performance [2-7]. More specifically, two-device structures have been employed in OTFTs, a bottom and a top contact. It has been well known that top-contact (TC) devices show better performance than bottom-contact (BC) devices, since low contact resistance between source/drain (S/D) electrodes and a semiconductor layer allows less electron-scattering at the interface [8]. Moreover, band alignment of the electrodes and semiconductor layers has become the key to attaining high device performance. For example, NiO_x and RuO_x prepared by sputtering process have more suitable work func-

tion (Φ) than pure metals for *p*-type organic semiconductors. In a BC-pentacene TFTs, the higher field-effect mobility and on/off ratio were observed when NiO_x and RuO_x ($\Phi=5.0$ eV and 4.92 eV, respectively) were used, which was attributed to the more efficient hole injection from the highest occupied molecular orbital (HOMO, 5.1 eV) of pentacene to the metal electrodes than when Ni ($\Phi=4.6$ eV), Ru ($\Phi=4.64$ eV) and even Au ($\Phi=4.85$ eV) were used for electrodes [9-11]. However, the effect of such metal oxides on organic semiconductor layers and device performance during the deposition has not been reported for TC-devices.

Herein, we report on the effect of radio frequency (RF) magnetron sputtering of RuO_x on pentacene thin films. After deposition of RuO_x, the oxidized pentacene, 6,13-pentacenequinone, is observed on the top surface of the pentacene film by X-ray photoelectron spectroscopy (XPS) analysis. Even though field-effect mobility of the devices is improved from $0.308 \text{ cm}^2/\text{V}\cdot\text{s}$ to $0.499 \text{ cm}^2/\text{V}\cdot\text{s}$ as the flow rate of oxygen gas increases, the high charge trap density at the pentacene/RuO_x interface leads to the high level of leakage currents and significant drop in the on/off ratio.

EXPERIMENTAL DETAILS

SiO₂ (300 nm)/N++Si substrates (Silicon Material Inc.) were immersed into 7:4 (volume ratio) H₂SO₄/H₂O₂ solution (piranha solution) at 100 °C for 1 h to remove organic residues, and cleaned in an UV tip-cleaner (UV TC 220, Pucotech) for 1 hr. To provide good surface coverage of pentacene on SiO₂, octadecyltrichlorosilane self-assembled monolayers (OTS-SAM) were deposited on the SiO₂ by immersing the substrates for 6 h at 4 °C into an anhydrous toluene (99.8%, Aldrich) solution containing 10 mmol of OTS (Aldrich)

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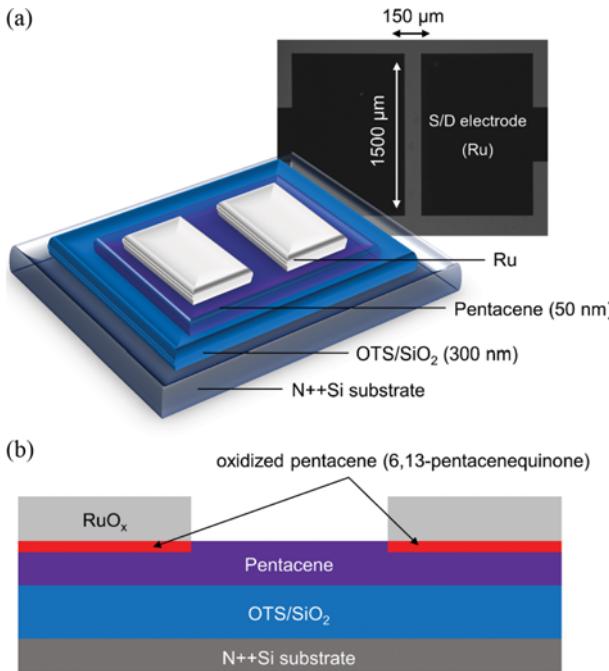


Fig. 1. Schematic illustrating (a) the pentacene TFT with a 150 μm -long and 1,500 μm -wide channel and (b) the leakage channel (indicated by the red squares) formed at the pentacene/electrode interface.

[12]. The substrates treated with SAM were further cleaned in acetone, and annealed in a vacuum chamber for 2 h at 150 $^{\circ}\text{C}$. 50 nm-thick pentacene thin films were deposited in an organic molecular

beam deposition (OMBD) reactor at 70 $^{\circ}\text{C}$ and 2×10^{-6} Torr with a deposition rate of 0.2–0.3 \AA/sec . A shadow mask was used during the deposition to pattern the pentacene layer. The Ru and RuO_x S/D electrodes were deposited with RF magnetron sputtering under the argon and oxygen atmosphere, respectively. The deposition was at a working pressure of 7 mTorr with the RF power of 50 W. The composition of the sputtering atmosphere was controlled by controlling the flow rate of argon and oxygen. The channel dimension had a length of 150 μm and a width of 1,500 μm as shown in Fig. 1(a). The I-V characteristics of the device were measured with an Agilent E5270A precision semiconductor parameter analyzer. Oxidation of pentacene thin films was by using an UV tip-cleaner with different exposure times (30, 60, 120 and 180 min). The surface morphology of pentacene was investigated with an atomic force microscope (AFM, Veeco Dim. V). The variations in the molecular structure of pentacene were observed with a X-ray photoelectron spectroscope (XPS, the 4B1 beam line in the Pohang Acceleration Laboratory synchrotron radiation source).

RESULTS AND DISCUSSION

Fig. 1(a) (top view) and 1(b) (side view) exhibit the schematic of the TC-pentacene TFT. When RuO_x electrodes were deposited in the presence of oxygen plasma, highly reactive oxygen species could react with the pentacene film [13]. As shown in Fig. 1(b), such reaction has an effect right on the top surface of the pentacene film, and pentacenequinone derivatives are formed. To elucidate the effect of oxygen plasma damage, the performance of both BC- and TC-devices, and the surface morphologies of pentacene should be investigated.

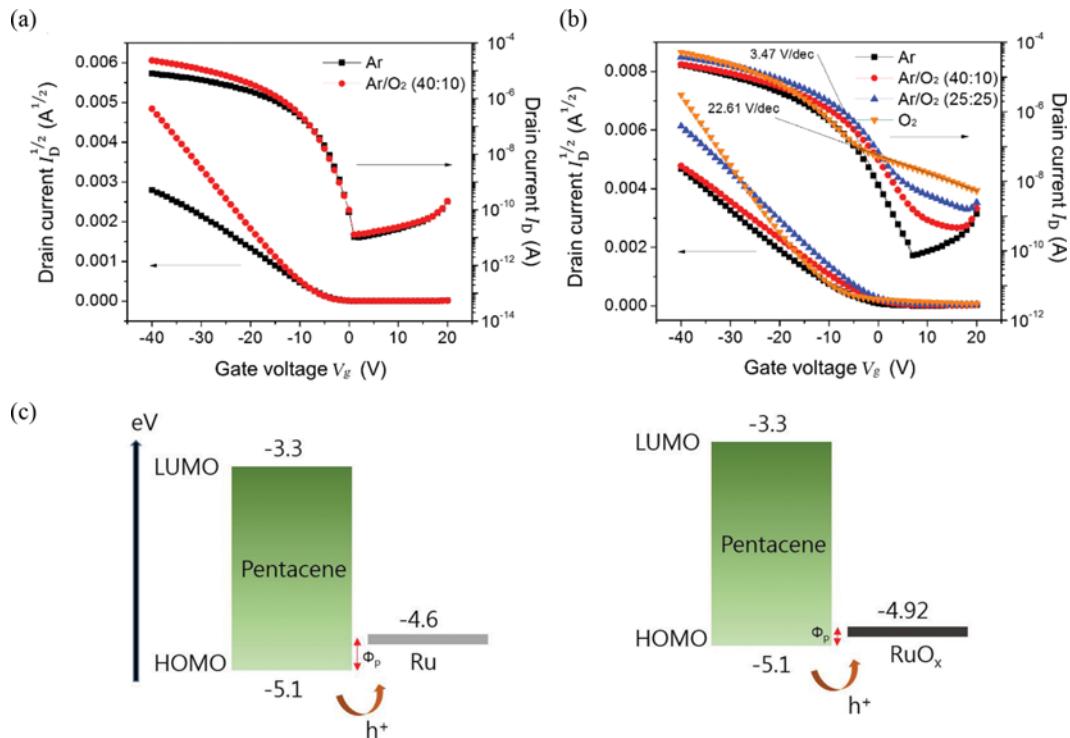


Fig. 2. Saturation transfer curves of the (a) BC- and (b) TC-devices. (c) Schematic illustrating energy diagram of pentacene, Ru and RuO_x.

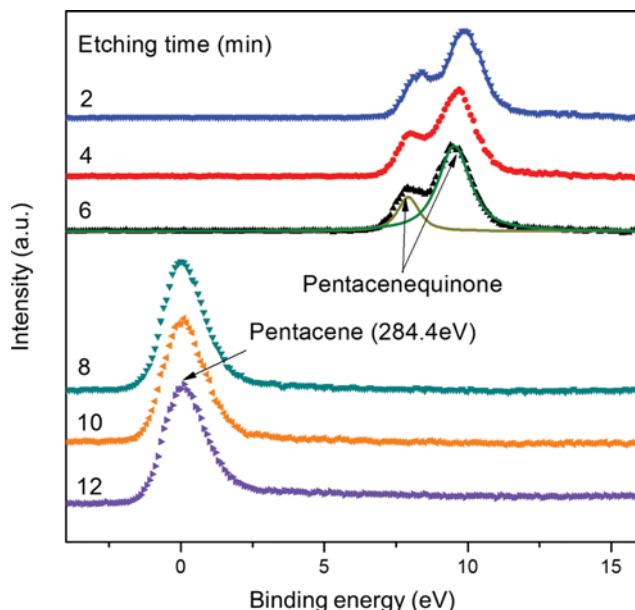


Fig. 3. C1s XPS spectra (depth profile) of the 50 nm-thick pentacene film exposed to the oxygen plasma for 30 s.

plasma for 30 s. During Ar etching of a 50 nm-thick pentacene thin film (etching rate=1 nm/min), the characteristic peaks of 6, 13-pentacenequinone were observed until 6 min. It was confirmed by the emergence of two peaks from single peak (284.4 eV) at the C1s core level corresponding to 284.9 eV and 287.3 eV, respectively [20, 21]. The peaks disappeared after further etching of the thin film (etching time>8 min). That is, about 6 nm-thick 6, 13-pentacenequinone layer was created from the outermost surface of pentacene. This demonstrated that the generation of C=O and C-O bonds in 6, 13-pentacenequinone by oxidation resulted in the charge trapping at the interface with electrodes, which produced the high level of leakage currents at the interface, not at the bottom layer of pentacene contacted to SiO₂.

We also investigated the morphological changes in the pentacene films by using AFM. Fig. 4 shows the AFM images obtained after oxygen plasma and ultraviolet (UV) light treatment to the films with different exposure times (0-180 min). In addition to oxygen plasma, UV light also generates highly reactive oxygen radicals in air and they oxidize organic materials [22-24]. The pentacene grains (200-

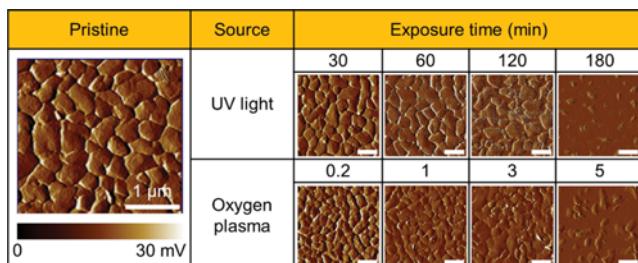


Fig. 4. AFM images (5 $\mu\text{m} \times 5 \mu\text{m}$) of the 50 nm-thick pentacene films obtained by tapping mode. The samples were exposed to UV light and oxygen plasma for different exposure times. The scale bars are 500 nm.

500 nm) were easily decomposed and evaporated at room temperature, since 6, 13-pentacenequinone consisted of an intermediate phase, and it was volatile as well [25]. The fast degradation speed under oxygen plasma treatment was attributed to the fact that higher power of oxygen plasma (50 W) than that of UV light (20 W) produced a higher level of oxygen radicals. From these results, it should be considered that the sputtering process to deposit metal oxide-electrodes results in severe damage to an organic semiconductor at the configuration of TC-devices.

CONCLUSIONS

We observed that RF magnetron sputtering of Ru damaged pentacene thin films and degraded their electrical properties. The oxidation of the pentacene film by oxygen plasma gave rise to many charge traps at the pentacene/electrode interface. As the result, the subthreshold slopes and leakage currents in the TC-RuO_x devices significantly increased, whereas the performance of BC-RuO_x devices was superior to that of BC-Ru devices. From the XPS analysis, the oxidized pentacene film exhibited a significant positive shift of binding energy due to its high oxidation level, which turned out to be 6, 13-pentacenequinone derivatives. The AFM images showed that pentacene films were readily decomposed in the presence of both oxygen plasma and UV light. These results suggest that the chemical damage to organic semiconductors by the reactive gas species should be considered when TC-OTFTs are fabricated.

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REFERENCES

- K. T. Kamtekar, A. P. Monkman and M. R. Bryce, *Adv. Mater.*, **22**, 572 (2010).
- C. R. Newman, C. D. Frisbie, D. A. da Silva Filho, J.-L. Bredas, P.C. Ewbank and K.R. Mann, *Chem. Mater.*, **16**, 4436 (2004).
- E. Fortunato, P. Barquinha and R. Martins, *Adv. Mater.*, **24**, 2945 (2012).
- A. R. Murphy and J. M. J. Fréchet, *Chem. Rev.*, **107**, 1066 (2007).
- S. W. Rhee and D. J. Yun, *J. Mater. Chem.*, **18**, 5437 (2008).
- A. Facchetti, M. H. Yoon and T. J. Marks, *Adv. Mater.*, **17**, 1705 (2005).
- C. D. Dimitrakopoulos and P. R. L. Malenfant, *Adv. Mater.*, **14**, 99 (2002).
- P. Cosseddu and A. Bonfiglio, *Thin Solid Films*, **515**, 7551 (2007).
- D. J. Yun, S. Lee, K. Yong and S.-W. Rhee, *Appl. Phys. Lett.*, **97**, 073303 (2010).
- D. J. Yun and S. W. Rhee, *J. Electrochem. Soc.*, **155**, H899 (2008).
- C. W. Chu, S. H. Li, C. W. Chem, V. Shrotriya and Y. Yang, *Appl. Phys. Lett.*, **87**, 193508 (2005).

12. X.-H. Zhang, B. Domercq, X. Wang, S. Yoo, T. Kondo, Z. L. Wang and B. Kippelen, *Org. Electron.*, **8**, 718 (2007).
13. D. H. Kim, D. W. Kim, K. S. Kim, H. J. Kim, J. S. Moon, M. P. Hong, B. S. Kim, J. H. Shin, Y. M. Kim, K. K. Song and S. S. Shin, *Jpn. J. Appl. Phys.*, **47**, 5672 (2008).
14. A. Rolland, J. Richard, J.-P. Kleider and D. Mencaraglia, *J. Electrochem. Soc.*, **140**, 3679 (1993).
15. M. McDowell and I. G. Hill, *Appl. Phys. Lett.*, **88**, 073505 (2006).
16. H. K. Kim, I.-H. Yu, J. H. Lee, T. J. Park and C. S. Hwang, *ACS Appl. Mater. Interfaces*, **5**, 1327 (2013).
17. S. Park, W. Kim and Y. Kim, *Korean J. Chem. Eng.*, **34**, 1500 (2017).
18. F. So and D. Kondakov, *Adv. Mater.*, **22**, 3762 (2010).
19. M. Park, J.-S. Park, I. K. Han and J. Y. Oh, *J. Mater. Chem. A.*, **4**, 11307 (2016).
20. S. J. Kang, Y. Yi, C. Y. Kim, K.-H. Yoo, A. Moewes, M. H. Cho, J. D. Denligner, C. N. Whang and G. S. Chang, *Phys. Rev. B*, **72**, 205328 (2005).
21. P. Parisse, S. Picozzi and L. Ottaviano, *Org. Elec.*, **8**, 498 (2007).
22. H. W. Zan and C.-W. Chou, *Jpn. J. Appl. Phys.*, **48**, 031501 (2009).
23. K. Ono, H. Totani, T. Hiei, A. Yoshino, K. Saito, K. Eguchi, M. Tomura, J. Nishida and Y. Yamashita, *Tetrahedron*, **63**, 9699 (2007).
24. Y. Matsumoto, T. Ohsawa, K. Nakajima and H. Koinuma, *Meas. Sci. Technol.*, **16**, 199 (2005).
25. A. Vollmer, H. Weiss, S. Rentenberger, I. Salzmann, J. P. Rabe and N. Koch, *Surf. Sci.*, **600**, 4004 (2006).